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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/072,201	02/07/2002	Hiroyuki Otaki	TJK/213	3764
26689	7590	03/26/2004	EXAMINER	
WILDMAN, HARROLD, ALLEN & DIXON 225 WEST WACKER DRIVE CHICAGO, IL 60606			ANGEBRANNNDT, MARTIN J	
			ART UNIT	PAPER NUMBER
			1756	

DATE MAILED: 03/26/2004

Please find below and/or attached an Office communication concerning this application or proceeding.

5/1

<b>Office Action Summary</b>	<b>Application No.</b>	<b>Applicant(s)</b>	
	10/072,201	OTAKI ET AL.	
	<b>Examiner</b>	<b>Art Unit</b>	
	Martin J Angebranndt	1756	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

#### Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If the period for reply specified above is less than thirty (30) days, a reply within the statutory minimum of thirty (30) days will be considered timely.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

#### Status

- 1) Responsive to communication(s) filed on 17 May 2003.
- 2a) This action is FINAL.                    2b) This action is non-final.
- 3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

#### Disposition of Claims

- 4) Claim(s) 1-34 is/are pending in the application.
- 4a) Of the above claim(s) \_\_\_\_\_ is/are withdrawn from consideration.
- 5) Claim(s) \_\_\_\_\_ is/are allowed.
- 6) Claim(s) 1-34 is/are rejected.
- 7) Claim(s) \_\_\_\_\_ is/are objected to.
- 8) Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement.

#### Application Papers

- 9) The specification is objected to by the Examiner.
- 10) The drawing(s) filed on \_\_\_\_\_ is/are: a) accepted or b) objected to by the Examiner.  
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).  
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

#### Priority under 35 U.S.C. § 119

- 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
  - a) All    b) Some \* c) None of:
  1. Certified copies of the priority documents have been received.
  2. Certified copies of the priority documents have been received in Application No. \_\_\_\_\_.
  3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

\* See the attached detailed Office action for a list of the certified copies not received.

#### Attachment(s)

- 1) Notice of References Cited (PTO-892)
- 2) Notice of Draftsperson's Patent Drawing Review (PTO-948)
- 3) Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08)  
Paper No(s)/Mail Date 5/17/2003.
- 4) Interview Summary (PTO-413)  
Paper No(s)/Mail Date. \_\_\_\_\_
- 5) Notice of Informal Patent Application (PTO-1447)
- 6) Other: \_\_\_\_\_

1. The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless—

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

(e) the invention was described in (1) an application for patent, published under section 122(b), by another filed in the United States before the invention by the applicant for patent or (2) a patent granted on an application for patent by another filed in the United States before the invention by the applicant for patent, except that an international application filed under the treaty defined in section 351(a) shall have the effects for purposes of this subsection of an application filed in the United States only if the international application designated the United States and was published under Article 21(2) of such treaty in the English language.

2. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

3. Claims 25,26,29,30 and 33-34 are rejected under 35 U.S.C. 102(b) as being fully anticipated by Nakamura et al. JP 05-346662.

Nakamura et al. JP 05-346662 (machine translation attached) teaches in example 1 [0132-0137] a photosensitive layer comprising a methacrylic acid/ allyl copolymer, a monomer, a silane coupling agent and a photoinitiator. Example 2 [0156-0164] a photosensitive layer comprising a methacrylic acid/ allyl copolymer, a monomer, a silane coupling agent and a two photoinititors. The examiner holds that photoinitiator c also acts as a sensitizing dye as it is a benzothiazole cyanine dye. Please note that these are not the toner layers/materials, but the photosensitive layer. The use of various spectral sensitizing agents including cyanine dyes is disclosed. [0039].

The carbonyl moieties of the binder are inherently held to chelate at least some of the silane coupling agent.

4. Claims 25,27,29,31 and 33-34 are rejected under 35 U.S.C. 102(b) as being fully anticipated by Sashida et al. '451.

Sashida et al. '451 describes in comparative examples 1 and 4, polymers having a number average MW of 8,000 or 6,000 daltons, Michler's ketone (photoinitiator), trimethoxysilylpropyl methacrylate (silane coupling agent) and diethylene glycol dimethacrylate as the monomer. (columns 9-11).

The carbonyl moieties of the binder are inherently held to chelate at least some of the silane coupling agent.

5. Claims 29-34 are rejected under 35 U.S.C. 103(a) as being unpatentable over Kawabata et al. '340.

Kawabata et al. '340 teach holographic recording compositions which include both cationically and free-radically polymerizable materials and are spin coated on a glass substrate. (see examples). The use of various additives including silane coupling agents is disclosed. (6/21). The use of cationically curable materials, such as the polyfunctional monomer disclosed at col 3/line 60- col. 4/line 3 is disclosed.

It would have been obvious to one skilled in the art to modify the examples by using other disclosed monomers, such as the polyfunctional monomer disclosed at col 3/line 60- col. 4/line 3 together with useful additives, such as the silane coupling agent would have been obvious and leads one to a reasonable expectation of success based upon the disclosed equivalent function and the increased adhesion to glass substrate being known as a result of adding coupling

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agents. The examiner holds that the language describing the binder includes higher MW epoxide monomers. The carbonyl moieties of the binder are inherently held to chelate at least some of the silane coupling agent.

6. Claims 29-34 are rejected under 35 U.S.C. 103(a) as being unpatentable over Kawabata et al. '340, in view of Kojima et al. '968.

Kojima et al. '968 establishes that silane coupling agents are known for their adhesion to glass substrates. (4/27-49)

In addition to the basis provided above, the examiner has cited Kojima et al. '968 to support the position that the resulting increased adhesion to the glass substrate due to the addition of a silane coupling agent is old and well known in the art and that it would have been obvious to one skilled in the art to add the silane coupling agent to gain increased adhesion of the polymerizable composition to the glass substrate used in the examples of Kawabata et al. '340.

7. Claims 25,26,29,30 and 33-34 are rejected under 35 U.S.C. 102(b) as being fully anticipated by Kikuchi JP 63-264686.

Kikuchi JP 63-264686 in examples 1 uses 40 parts methacrylic/styrene/maleic anhydride polymer, 20 parts of a trimethyolpropyltriacylate monomer, 10 parts of a pigment, 0.5 parts of a silane coupling agent and 10 parts of a hydroxyethyl acrylate monomer coated on a substrate. (page 6/upper right hand column). Example 2, uses an aluminum coupling agent (page 6/lower right hand column). Example 3 (upper left column on page 7) uses a titanium coupling agent.

The carbonyl moieties of the binder are inherently held to chelate at least some of the silane coupling agent.

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8. Claims 25-27,29-31 and 33-34 are rejected under 35 U.S.C. 102(b) as being fully anticipated by JP 06-161095 (machine translation attached).

See examples cited in abstract [0022-0023 in machine translation]. The examiner considered the light metal salt of the olefin maleic acid polymer to meet the binder requirements of the claims.

9. Claims 25-27,29-31 and 33-34 are rejected under 35 U.S.C. 102(b) as being fully anticipated by JP 2000-273418.

See example 1 in machine translation. The carbonyl moieties of the binder are inherently held to chelate at least some of the silane coupling agent.

10. Claims 25,26,28-30 and 32-34 are rejected under 35 U.S.C. 102(b) as being fully anticipated by JP 10-338850.

See example 1 in machine translation. The examiner holds that the language describing the binder includes higher MW epoxide (Epicoat 828)resin with the cationic initiator and are held to chelate the silane.

11. Claims 25-26,29-30 and 33-34 are rejected under 35 U.S.C. 102(b) as being fully anticipated by Mochel et al. '297.

See example (13/67-14/29, making polymer and 14/30-46, adding monomer, benzoin photoinitiator and aluminum chelating agent). The carbonyl moieties of the binder are inherently held to chelate at least some of the aluminum coupling agent.

12. Claims 25-27,29-31 and 33-34 are rejected under 35 U.S.C. 102(b) as being fully anticipated by Proskow '608.

See example 4, formulation A, which uses the Hycar CTBNW which has a Mw of 3,400.

The carbonyl moieties of the binder are inherently held to chelate at least some of the zinc coupling agent.

13. Claims 1,3-5,7-8,10-13,15-17, 19-23 are rejected under 35 U.S.C. 102(b) as being fully anticipated by Roos '306.

Roos '306 describes a photopolymerizable composition which includes a terpolymer of methacryloxypropyltrimethoxy silane, methyl methacrylate and methacrylonitrile, a trimethylolpropane triacrylate monomer, a chlorophenyl-4,5-bis(phenyl) imidazoyl dimer (initiator) and a coumarin sensitizer coated on a PET support. (3/70-4/22)

The organo-metallic particles are polymeric having at least one photopolymerization group. The examiner holds that the terpolymer of methacryloxypropyltrimethoxy silane, methyl methacrylate and methacrylonitrile meets both the hybrid polymer and the organometallic particle limitation. As polymerization was used to form the terpolymer, at least some of the groups would still be left on the polymer and it would be reactive and able to change its refractive index due to further curing by photopolymerization

14. Claims 1-24 are rejected under 35 U.S.C. 103(a) as being unpatentable over Maeda et al. '771 , in view of Sato et al. '846, Baney et al., "Silsesquioxanes" Chem. Rev. vol 95(5) pp. 1409-1430 and Krug et al., "Fine Patterning of Thin Sol-gel Films", J. non-cryst. Sol. Vol. 147/148 pp. 447-450, (1992).

Maeda et al. '771 teach the use of solgel processing to form useful holographic recording media, where a sol-gel processing is used to form a matrix in situ and mixed with photopolymerizable materials.

Sato et al. '846 teach the use of silicon containing monomeric compounds in holographic recording media, specifically those which contain both free radically and cationically polymerizable materials. (see examples) Useful silicon containing cationically curable compounds are disclosed (3/48-4/4). The use of cationically reactive binders is disclosed. (7/15-24). The use of silane coupling agents is also disclosed. (7/11)

Baney et al., "Silsesquioxanes" Chem. Rev. vol 95(5) pp. 1409-1430 describes various techniques for forming organic/inorganic hybrid, including those containing moieties which may be photocured, such as epoxides and vinyl moieties. (page 1426, section VB, right column)

Krug et al., "Fine Patterning of Thin Sol-gel Films", J. non-cryst. Sol. Vol. 147/148 pp. 447-450, (1992) teaches the method for forming photocurable solo-gel polymers where methacrylate monomers are reacted with alkoxides and then with other monomers.

15. It would have been obvious to modify the process of Maeda et al. '771 by using producing a reactive sol-gel matrix containing epoxide moieties based upon the disclosure within Sato et al. '846 that these would be compatible and that epoxide containing reactive binder are desirable in holographic recording media, using technique similar to those disclosed by Baney et al., "Silsesquioxanes" Chem. Rev. vol 95(5) pp. 1409-1430 and Krug et al., "Fine Patterning of Thin Sol-gel Films", J. non-cryst. Sol. Vol. 147/148 pp. 447-450, (1992) to form the epoxide containing polymerizable matrix with a reasonable expectation of forming a useful holographic recording medium.

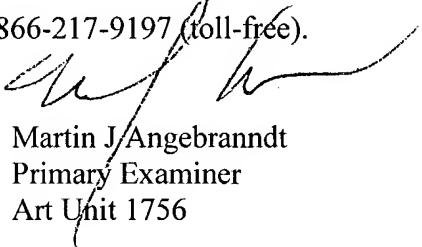
16. The prior art made of record and not relied upon is considered pertinent to applicant's disclosure.

Bauer et al. EP 243933 (PTO-1449) see example 1 is considered cumulative

17 Any inquiry concerning this communication or earlier communications from the examiner should be directed to Martin J Angebranndt whose telephone number is 571-272-1378. The examiner can normally be reached on Monday-Thursday and alternate Fridays.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Mark Huff can be reached on 571-272-1385. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).



Martin J Angebranndt  
Primary Examiner  
Art Unit 1756

03/22/2004